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Key indicators

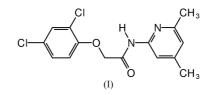
Single-crystal X-ray study T = 220 KMean σ (C–C) = 0.004 Å R factor = 0.039 wR factor = 0.109 Data-to-parameter ratio = 13.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. 2-(2,4-Dichlorophenoxy)-*N*-(4,6-dimethylpyridin-2-yl)acetamide

The structure of the title compound, $C_{15}H_{14}Cl_2N_2O_2$, comprises an essentially flat molecule [dihedral angle of 3.64 (9)° between the two aromatic rings] with an intramolecular hydrogen-bonding interaction from the amine N–H group to the phenoxy O atom. Molecules are arranged in lamellar sheets parallel to the (110) plane. Received 16 July 2003 Accepted 24 July 2003 Online 15 August 2003

Comment

A search of the April 2003 release of the Cambridge Structural Database (Allen, 2002) reveals that there are eight structures based on N-(4,6-dimethylpyridin-2-yl)carboxamide, all published by one research group. These structures are N-(4,6-dimethylpyridin-2-yl)-2-(3-nitrophenyl)acetamide (Rodier et al., 1986), N-(4,6-dimethylpyridin-2-yl)-2-(4-nitrophenyl)propionamide (Rodier, Robert & Le Baut, 1990), (E)-N-(4,6-dimethylpyridin-2-yl)-3-phenylpropenamide hemihydrate (Rodier, Robert-Piessard & Le Baut, 1990), N-(4,6dimethylpyridin-2-yl)-2-furamide (Rodier et al., 1991), N-(4,6dimethylpyridin-2-yl)-2-thiophenecarboxamide (Rodier, Cense et al., 1992), N-(4,6-dimethylpyridin-2-yl)(1-methylindol-2-yl)carboxamide (Rodier, Robert & Le Baut, 1992), N-(4,6-dimethylpyridin-2-yl)-2-(3-nitrophenyl)acetamide (Rodier et al., 1993) and N-(4,6-dimethylpyridin-2-yl)-5-methylpyrazine-2carboxamide (Rodier et al., 1994). In a series of studies on the syntheses of additional derivatives of N-(4,6-dimethylpyridin-2yl)carboxamide and also N-(4,6-dimethylpyrimidin-2-yl)carboxamide as potential anti-inflammatory agents, we prepared the title compound, (I), and its structure is reported here.



The structure of (I) comprises an essentially flat molecule (Fig. 1) with an intramolecular hydrogen-bonding interaction from the amine N-H group to the phenoxy O atom (Table 1). Molecules of (I) are arranged in lamellar sheets parallel to the (110) plane. The dihedral angle between the substituted phenyl and pyridyl rings is 3.64 (9)°.

Experimental

Four molar equivalents of oxalyl chloride (2.29 g, 18.1 mmol) were added dropwise to a stirred solution of (2,4-dichlorophenoxy)acetic acid (1.0 g, 4.5 mmol) and a catalytic amount of dry pyridine (dried over KOH) in 20 ml dry dichloromethane (dried over molecular sieves). After stirring for 30 min, the solvent was removed under

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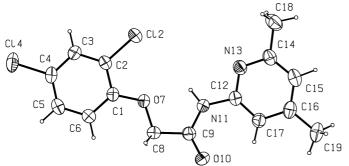


Figure 1

The molecular configuration and atom-numbering scheme for the title compound, showing 50% probability displacement ellipsoids.

reduced pressure, yielding a yellow solution of (2,4-dichlorophenoxy) acetic chloride. This solution was then added dropwise to a stirred solution of 2-amino-4,6-dimethyl pyridine (0.55 g, 4.5 mmol) and triethylamine (0.46 g, 4.5 mmol) in 20 ml dry dichloromethane. After 30 min, the solution was filtered and the solvent removed under reduced pressure, yielding a white powder. The product was purified using column chromatography (SiO₂) and collected in the eluted ethyl acetate fraction after initial chloroform elutions. The solid product was further washed with 30 ml cold ethanol to afford 0.82 g of (I) (81%); m.p. 435–436 K; ¹H NMR (CDCl₃, 250 MHz): δ (p.p.m.) 2.33 (s, 3H, CH₃), 2.42 (s, 3H, CH₃), 4.62 (s, 2H, CH₂), 6.77 (s, 1H, Ar–H), 6.89 (d, J = 8.8 Hz, 1H, Ar–H), 7.33 (d, J = 8.8 Hz, 1H, Ar–H), 7.41 (d, J = 2.5 Hz, 1H, py–H), 7.87 (s, 1H, py–H), 8.89 (bs, 1H, NH); m/z 325 (MH⁺, 100%), 327 (MH⁺, 75%). Crystals suitable for X-ray diffraction studies were grown from ethyl acetate.

Crystal data

 $\begin{array}{l} C_{15}H_{14}Cl_2N_2O_2\\ M_r = 325.18\\ Triclinic, P\overline{1}\\ a = 7.445~(5)~\text{\AA}\\ b = 8.790~(6)~\text{\AA}\\ c = 12.549~(7)~\text{\AA}\\ a \approx 89.28~(4)^\circ\\ \beta = 87.08~(5)^\circ\\ \gamma = 68.74~(4)^\circ\\ V = 764.4~(9)~\text{\AA}^3 \end{array}$

Data collection

Stoe Stadi-4 diffractometer ω/θ scans 3678 measured reflections 2570 independent reflections 2032 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.019$ $\theta_{\text{max}} = 25.0^{\circ}$ Z = 2 $D_x = 1.413 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 22 reflections $\theta = 15-16^{\circ}$ $\mu = 0.43 \text{ mm}^{-1}$ T = 220 (2) KPrism, colourless $0.70 \times 0.47 \times 0.38 \text{ mm}$

 $h = -8 \rightarrow 8$ $k = -10 \rightarrow 10$ $l = 0 \rightarrow 14$ 3 standard reflections frequency: 60 min intensity decay: 3% Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.062P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.039$	+ 0.1634P]
$wR(F^2) = 0.109$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.05	$(\Delta/\sigma)_{\rm max} < 0.001$
2570 reflections	$\Delta \rho_{\rm max} = 0.20 \text{ e } \text{\AA}^{-3}$
196 parameters	$\Delta \rho_{\rm min} = -0.30 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1 Hydroge

Hydrogen-bonding geometry (A,	°)	•
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$D - H \cdots A$	<i>D</i> -H	$H \cdots A$	$D \cdots A$	$D - H \cdots A$
N11-H11···O7	0.87	2.07	2.552 (3)	114

All H atoms were included in the refinement at calculated positions, as riding atoms, with N–H set to 0.87 Å and C–H set to 0.97 (CH₃), 0.98 (CH₂) or 0.94 Å (Ar–H); the isotropic displacement parameters were set at 1.25 times U_{eq} of the carrier atom.

Data collection: *DIF*4 (Stoe & Cie, 1990); cell refinement: *DIF*4; data reduction: *REDU*4 (Stoe & Cie, 1990); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *PLATON*97 (Spek, 1997); software used to prepare material for publication: *SHELXL*97.

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References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Rodier, N., Cense, J. M., Robert, J.-M. & Le Baut, G. (1991). Acta Cryst. C47, 2688–2690.
- Rodier, N., Cense, J. M., Robert, J.-M. & Le Baut, G. (1992). Acta Cryst. C48, 1148–1150.
- Rodier, N., Piessard, S. & Le Baut, G. (1986). Bull. Soc. Chim. Fr. pp. 418–422.Rodier, N., Rideau, O., Robert, J.-M. & Le Baut, G. (1994). Acta Cryst. C50, 1760–1762.
- Rodier, N., Robert, J.-M. & Le Baut, G. (1990). Acta Cryst. C46, 154-156.
- Rodier, N., Robert, J.-M. & Le Baut, G. (1992). Acta Cryst. C48, 572-574.
- Rodier, N., Robert, J.-M., Robert-Piessard, S. & Le Baut, G. (1993). Acta Cryst. C49, 154–156.
- Rodier, N., Robert-Piessard, S. & Le Baut, G. (1990). Acta Cryst. C46, 1747– 1749.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (1997). PLATON97. University of Utrecht, The Netherlands.
- Stoe & Cie (1990). DIF4 and REDU4. Stoe & Cie GmbH, Darmstadt, Germany.